

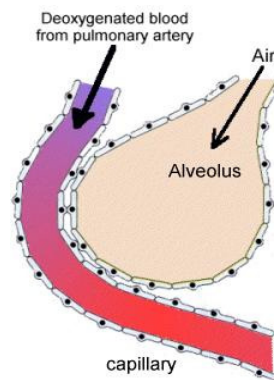
## AN INTRODUCTION TO GAS EXCHANGE

### Objectives

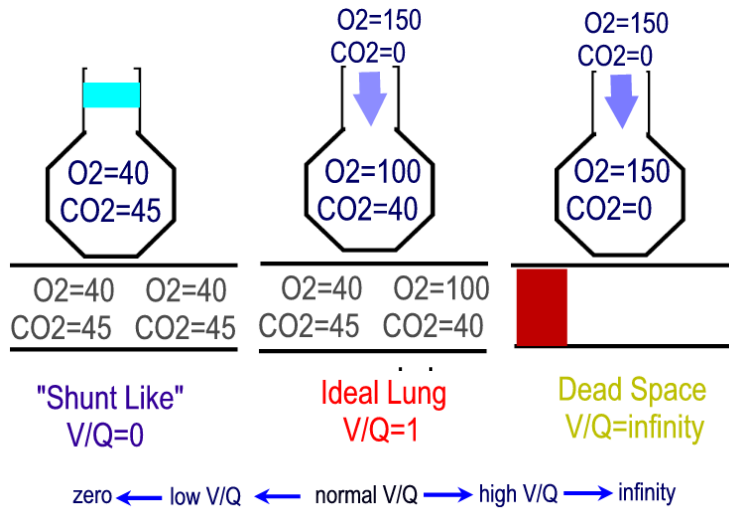
1. Define the partial pressure of a gas and explain why  $PO_2$  in is reduced as atmospheric air enters the airways and reaches the alveoli.
2. Compare the  $PO_2$  &  $PCO_2$  the blood in the pulmonary artery entering (mixed venous blood) entering the pulmonary capillaries the  $PO_2$  &  $PCO_2$  the arterial blood in the pulmonary veins leaving the pulmonary capillaries.
3. Describe the factors that affect the rate of exchange of  $O_2$  &  $CO_2$  across the alveolar-capillary membrane and give examples of disease states where these are altered.
4. Discover the partial pressure gradients that drive the diffusion of  $O_2$  &  $CO_2$  across the systemic capillaries and specify how they vary depending on the metabolic activity of the cells.

## PARTIAL PRESSURE OF ALVEOLAR GASES IS REFLECTED IN THE ARTERIAL BLOOD

Gas exchange results in end capillary blood gases that are in equilibrium with the partial pressure of gases in the alveoli

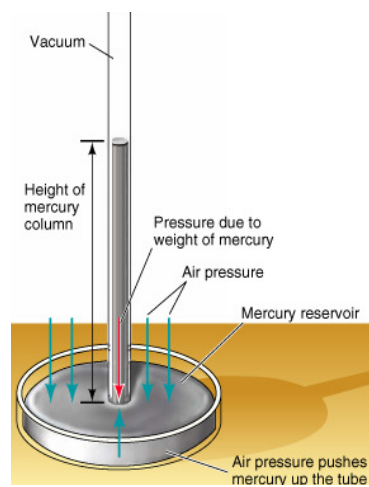


## VENTILATION PERFUSION MISMATCH AFFECTS ARTERIAL BLOOD GAS VALUES



## ATMOSPHERIC or BAROMETRIC PRESSURE [ $P_B$ ]

- Dry air is composed of a mixture of gases.
- Each gas exerts a **partial pressure** which is the pressure it would exert if it alone occupied a given volume.
- Barometric pressure [ $P_B$ ] is the total pressure exerted by this mixture of gases = 760 mmHg at sea level



## COMPOSITION OF GASES IN ATMOSPHERIC AIR Determining Their Partial Pressure

Nitrogen	78.08
Oxygen	20.94
Argon	0.93
Carbon Dioxide	0.03

and trace amounts of--  
Neon  
Methane  
Helium  
Krypton  
Hydrogen  
Xenon

Total atmospheric pressure = 760 mm Hg

Composition and partial pressures in atmospheric air

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Partial pressure of  $N_2$  in atmospheric air:  
 $P_{N_2} = 760 \text{ mm Hg} \times 0.79 = 600 \text{ mm Hg}$

Partial pressure of  $O_2$  in atmospheric air:  
 $P_{O_2} = 760 \text{ mm Hg} \times 0.21 = 160 \text{ mm Hg}$

## DETERMINING THE PARTIAL PRESSURE OF A GAS in a mixture of gases & water vapour

$$P_{\text{gas}} = P_B \times F_{\text{gas}}$$

In dry air at sea level

$$PO_2 = 760 \text{ mmHg} \times 0.21 = 160 \text{ mmHg}$$

Inspired air is heated to 37°C, humidified & saturated with water vapour  $PH_2O = 47 \text{ mmHg}$

During inspiration in the conducting airways

$$PO_2 = [760 - 47 \text{ mmHg}] \times 0.21 = 150 \text{ mmHg}$$

THE PARTIAL PRESSURE OF OXYGEN DECREASES AS  
INSPIRED AIR COURSES THROUGH THE AIRWAYS

PO<sub>2</sub> = 160 mmHg  
PCO<sub>2</sub> = 0  
P<sub>H2O</sub> = 0

DRY AIR AT SEA LEVEL

PO<sub>2</sub> = 150  
PCO<sub>2</sub> = 0  
P<sub>H2O</sub> = 47

CONDUCTING AIRWAYS  
(HEATED & HUMIDIFIED)

PO<sub>2</sub> = 100  
PCO<sub>2</sub> = 40  
P<sub>H2O</sub> = 47

ALVEOLI  
(MIXED IN A RESERVOIR OF AIR--FRC)

DETERMINE THE PARTIAL PRESSURE OF OXYGEN  
AT HIGH ALTITUDE

At the summit of Mount Everest at 18,000 ft, barometric pressure is half of its value at sea level.

What happens to the partial pressure of inspired oxygen?  
How about the PO<sub>2</sub> at the level of carina?

At the summit:

$$PO_2 = [760/2] \times 0.21 = 380 \times 0.21 = 80 \text{ mmHg}$$

In the carina:

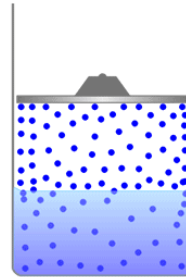
$$PIO_2 = [380 - 47] \times 0.21 = 333 \times 0.21 = 70 \text{ mmHg}$$

In the alveoli ?

## PARTIAL PRESSURE OF GASES DISSOLVED IN SOLUTION

At equilibrium, the tension of gas in solution = the partial pressure of gas in the pocket above it.

- Gases flow from a region of higher partial pressure to one of lower pressure.
- Gases dissolve, diffuse, and react according to their partial pressures and not according to their concentration in solution



note that the concentration of gas in solution is not the same as its partial pressure as solubility of gas affects its concentration

## A Key Factor In The Amount Of Gas Exchange Is The Partial Pressure Difference Across The Gas Exchange Barrier a.k.a. the driving pressure

### Across pulmonary capillaries

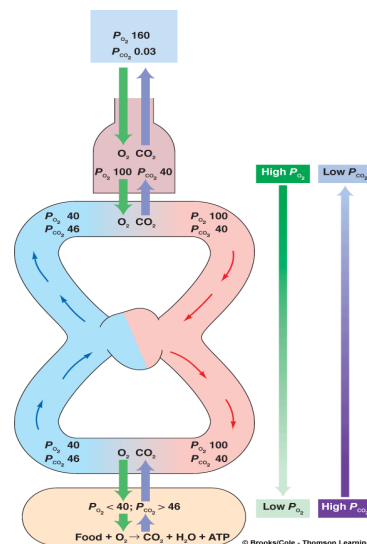
O<sub>2</sub> partial pressure gradient from alveoli to blood = 60 mm Hg (100 → 40)

CO<sub>2</sub> partial pressure gradient from blood to alveoli = 6 mm Hg (46 → 40)

### Across tissue capillaries

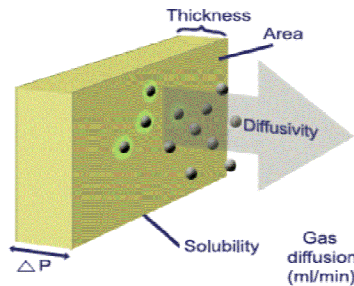
O<sub>2</sub> partial pressure gradient from blood to tissue = 60 mm Hg (100 → 40)

CO<sub>2</sub> partial pressure gradient from tissue cell to blood = 6 mm Hg (46 → 40)



## FACTORS THAT AFFECT DIFFUSION OF GASES ACROSS BARRIERS

Fick's Law Of Diffusion



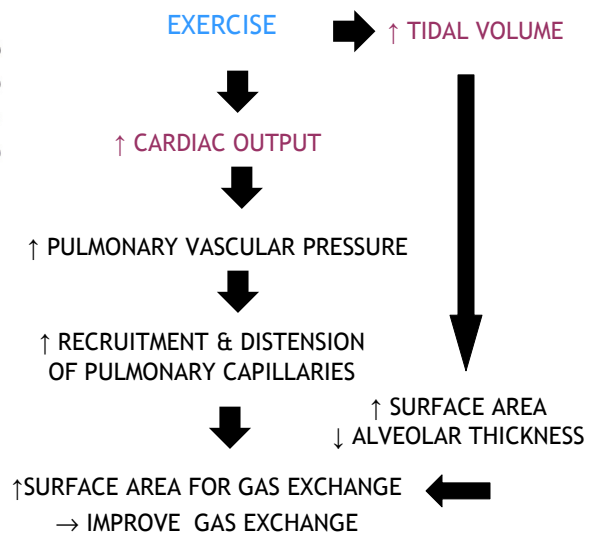
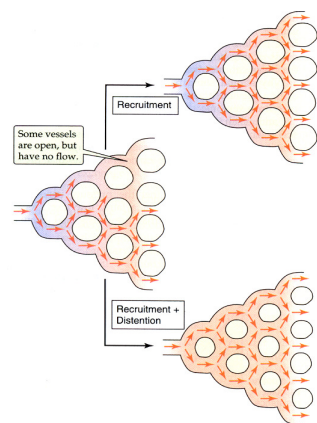
$$\dot{V}_{\text{GAS}} \propto \frac{A D [\Delta P]}{T}$$

$D$  = "DIFFUSIVITY" or "DIFFUSION COEFFICIENT"  $\propto$  SOLUBILITY /  $\sqrt{MW}$

Compare  $D$  &  $\Delta P$  of  $O_2$  and  $CO_2$  and explain equal time for end capillary equilibration of these gases in the lungs.

## EFFECT OF SURFACE AREA ON GAS EXCHANGE

Increase in Surface Area



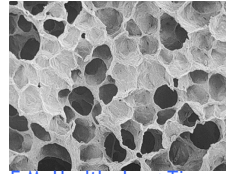
## EFFECT OF SURFACE AREA ON GAS EXCHANGE

Loss of Surface Area

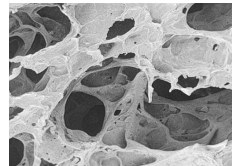
### EMPHYSEMA

↓  
DESTRUCTION OF ALVEOLAR WALLS

↓  
↑LUNG COMPLIANCE  
MECHANICAL COMPLICATION  
&  
AREA FOR GAS EXCHANGE  
↓GAS EXCHANGE COMPLICATION



E.M. Healthy Lung Tissue



E.M. Emphysematous Lungs

### Other

- Athelectasis (Collapse Of Lung Parenchyma)
- Surgical Removal Of Lung Tissue

## EFFECT OF THICKNESS ON GAS EXCHANGE

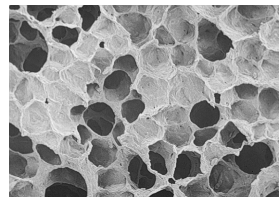
Increased Thickness

### PULMONARY FIBROSIS

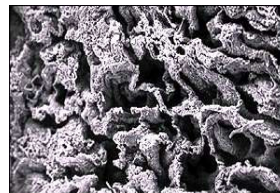
↓  
COLLAGEN DEPOSITION & SCAR  
FORMATION IN ALVEOLAR WALLS

↓  
MECHANICAL COMPLICATION  
↓ LUNG COMPLIANCE

GAS EXCHANGE COMPLICATION  
↑THICKNESS FOR GAS EXCHANGE



E.M. Healthy Lung Tissue



E.M. Fibrotic Lung Tissue

## EFFECT OF THICKNESS ON GAS EXCHANGE

### Increased Thickness

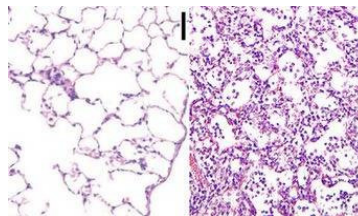
**PULMONARY FIBROSIS** (scar formation)  
results in response to exposure to--

- industrial dust [asbestosis, silicosis]
- spores from moldy hay [Farmers lungs]
- antigens in avian feather/excreta [Bird breeders lung]
- therapeutic drugs, radiation, poisons [weed killer, paraquat]

## EFFECT OF THICKNESS ON GAS EXCHANGE

### Increased Thickness

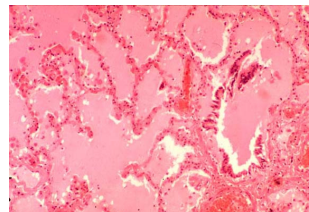
- **Pneumonia**- inflammatory fluid accumulation in & around the alveoli from viral/bacterial (common: *Streptococcal pneumonia*- deadly: *Severe Acute Respiratory Syndrome (SARS)*-the alveolar infection can reach blood--sepsis or the pleural space--empyema)  
/fungal infections or accidental aspiration of food, vomitus, chemicals or lung injury or indirectly as result of other illness.  
Pneumonia is the leading cause of death amongst chronically ill elderly



normal

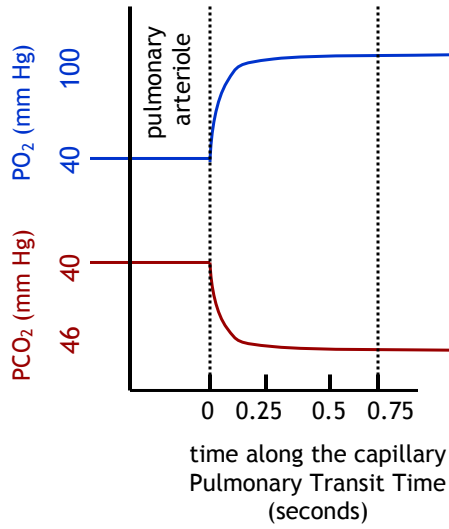
pneumonia

- **Pulmonary Edema**  
abnormal build up of fluid in the interstitial or alveolar space as a result of direct damage to lung tissue or inability of heart to pump blood --as in left heart failure



pulmonary edema

## GAS EXCHANGE ACROSS THE PULMONARY CAPILLARY Is Complete Within 1/4 Second



- at rest pulmonary transit time [ $\frac{3}{4}$  second] is more than that required to complete gas exchange [ $\frac{1}{4}$  second].
- during exercise, despite increased cardiac output, pulmonary transit time remains  $> \frac{1}{4}$  second & gas exchange is complete.
- in pulmonary fibrosis, reduced gas exchange is often seen in patients during exercise. At rest, the additional time spent in the capillary is sufficient to compensate for the thickened barrier .
- elite athletes with very high cardiac outputs have pulmonary transit times below  $\frac{1}{4}$  second during intense exercise  $\rightarrow$  inadequate oxygen exchange at the lungs  $\rightarrow$  low oxygen levels in the blood [arterial hypoxemia]